

mobius

Logarithmic Scales - Magnitude to Measured Value (Number)



What is the hydrogen ion concentration of a solution with a pH of 8 on the pH scale?

$$pH = -\log [H^+]$$
$$pH = 8$$

What is the hydrogen ion concentration of a solution with a pH of 6 on the pH scale?

$$pH = -\log [H^+]$$

 $pH = 6$

$$^{^{\hat{}}}\left[\mathsf{H}^{+}
ight]=1 imes10^{-8}\mathsf{mL/mol}$$

$$^{\hat{}}\left[\mathsf{H}^{+}
ight]=1 imes10^{-8}\mathsf{mL/mol}\,\left[\left[\mathsf{H}^{+}
ight]=3.16 imes10^{-5}\mathsf{mL/mol}
ight]$$

$$ilde{[}\mathsf{H}^{+}]=3.16 imes10^{-7}\mathsf{mL/mol}$$

$$[H^+] = 3.16 imes 10^{-7} \mathrm{mL/mol}^{\mathrm{B}} [H^+] = 1 imes 10^{-6} \mathrm{mL/mol}^{\mathrm{B}}$$

3 What is the hydrogen ion concentration of a solution with a pH of 9 on the pH

$$\mathsf{pH} = -\log\left[\mathsf{H}^+
ight] \ \mathsf{pH} = 9$$

What is the hydrogen ion concentration of a solution with a pH of 1 on the pH scale?

$$\mathsf{pH} = -\log\left[\mathsf{H}^+
ight] \ \mathsf{pH} = 1$$

$$^{^{\scriptscriptstyle{\mathsf{A}}}}[\mathsf{H}^{+}] = 1 imes 10^{-9} \mathsf{mL/mol}$$

$$[\mathsf{H}^+] = 0.1 \mathsf{mL/mol}$$

$${}^{ ilde{ iny b}}[\mathsf{H}^+] = 1 imes 10^{-8} \mathsf{mL/mol}$$

$$[\mathsf{H}^+] = 0.001 \mathrm{mL/mol}$$

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$$\mathsf{dB} = \mathsf{10} \, \mathsf{log} \, (\frac{\mathsf{I}}{\mathsf{I}_0})^{\mathsf{What is the sound intensity of a sound with a sound intensity of 20 dB}}$$

$$\mathsf{dB} = 10\log{(rac{\mathsf{I}}{\mathsf{I}_0})}$$

$$I_0 = 10^{-12} \text{W/m}^2$$
 $I_0 = 10^{-12} \text{W/m}^2$ $I_0 = 10^{-12} \text{W/m}^2$

$$m h=1\times10^{-8}W/m^2$$

$$I_0 = 10^{-12} \text{W/m}^2$$

$$\int_{0}^{1} = 1 \times 10^{-6} \text{W/m}^2$$

$$\beta = 20 \mathrm{dB}$$

$${f l}^{\scriptscriptstyle B}=1 imes 10^{-10}{f W/m^2}$$

$$m I = 1 \times 10^{-5} W/m^2$$

 $M = \log ($

$$\mathsf{dB} = 10\log{(rac{\mathsf{I}}{\mathsf{I}_0})}$$

$${f I}_0=1\mu{f m} \ {f M}=3$$

$${f l}=10\mu{
m m}$$
 ${f l}=1$, 000 $\mu{
m m}$

$$egin{aligned} \mathsf{I}_0 &= \mathsf{10}^{-12} \mathsf{W/m}^2 \ eta &= \mathsf{80dB} \end{aligned} \hat{\mathsf{I}} = 0.0001 \mathsf{W/m}^2 \end{aligned}$$

$$\hat{\mathsf{l}} = 0.0001 \mathrm{W/m^2}$$

$$m \mathring{l}=0.01W/m^2$$

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