



Logarithmic Scales - Magnitude to Measured Value (Power)



1

$$dB = 10 \log \left(\frac{I}{I_0} \right)$$

$$I_0 = 10^{-12} \text{ W/m}^2$$

$$\beta = 80 \text{ dB}$$

What is the sound intensity of a sound with a sound intensity of 80 dB on the decibel scale?

A

B

$$I = 10^{73} \text{ W/m}^2$$

$$I = 10^{68} \text{ W/m}^2$$

2

What is the hydrogen ion concentration of a solution with a pH of 4.7 on the pH scale?

$$\text{pH} = -\log [H^+]$$

$$\text{pH} = 4.7$$

A

$$[H^+] = 10^{-50,118} \text{ mL/mol}$$

B

$$[H^+] = 10^{-5,011,872} \text{ mL/mol}$$

3

$$M = \log \left(\frac{I}{I_0} \right)$$

$$I_0 = 1 \mu\text{m}$$

$$M = 2$$

What is the wave height of an earthquake with a magnitude of 2 on the Richter scale?

A

B

$$I = 10^{0.5} \mu\text{m}$$

$$I = 10^2 \mu\text{m}$$

4

What is the hydrogen ion concentration of a solution with a pH of 5.5 on the pH scale?

$$\text{pH} = -\log [H^+]$$

$$\text{pH} = 5.5$$

A

$$[H^+] = 10^{-316,227} \text{ mL/mol}$$

B

$$[H^+] = 10^{-3,162} \text{ mL/mol}$$

5

What is the hydrogen ion concentration of a solution with a pH of 5 on the pH scale?

$$\text{pH} = -\log [H^+]$$

$$\text{pH} = 5$$

$$[H^+] = 10^{-10,000,000} \text{ mL/mol}$$

$$[H^+] = 10^{-100,000} \text{ mL/mol}$$

6

$$dB = 10 \log \left(\frac{I}{I_0} \right)$$

$$I_0 = 10^{-12} \text{ W/m}^2$$

$$\beta = 101 \text{ dB}$$

What is the sound intensity of a sound with a sound intensity of 101 dB on the decibel scale?

A

B

$$I = 10^{94} \text{ W/m}^2$$

$$I = 10^{89} \text{ W/m}^2$$

7

$$M = \log \left(\frac{I}{I_0} \right)$$

$$I_0 = 1 \mu\text{m}$$

$$M = 6.9$$

What is the wave height of an earthquake with a magnitude of 6.9 on the Richter scale?

A

B

$$I = 10^{8.9} \mu\text{m}$$

$$I = 10^{6.9} \mu\text{m}$$

8

$$M = \log \left(\frac{I}{I_0} \right)$$

$$I_0 = 1 \mu\text{m}$$

$$M = 9.6$$

What is the wave height of an earthquake with a magnitude of 9.6 on the Richter scale?

A

B

$$I = 10^{9.6} \mu\text{m}$$

$$I = 10^{8.1} \mu\text{m}$$