



Logarithmic Scales - Ratio and Magnitude to Magnitude



1

$\text{pH} = -\log [\text{H}^+]$ If a solution has 1×10^9 times the Hydrogen ion concentration as one with a pH of 10 on the pH scale, what is its pH?

$\frac{[\text{H}^+]_2}{[\text{H}^+]_1} = 1 \times 10^9$

A	B
$\text{pH}_1 = 10$	$\text{pH}_2 = 2.5$ $\text{pH}_2 = 1$

2

$M = \log \left(\frac{I}{I_0} \right)$ If an earthquake has 1,000 times the wave size as one with a magnitude of 3 on the Richter scale, what is its magnitude?

$\frac{I_2}{I_1} = 1,000$

A	B
$M_1 = 3$	$M_2 = 6.5$ $M_2 = 6$

3

$\text{pH} = -\log [\text{H}^+]$ If a solution has 10 times the Hydrogen ion concentration as one with a pH of 10 on the pH scale, what is its pH?

$\frac{[\text{H}^+]_2}{[\text{H}^+]_1} = 10$

A	B
$\text{pH}_1 = 10$	$\text{pH}_2 = 11$ $\text{pH}_2 = 9$

4

$\text{dB} = 10 \log \left(\frac{I}{I_0} \right)$ If a sound has 1×10^{11} times the sound energy as one with a dB magnitude of 20 on the decibel scale, what is its dB magnitude?

$\frac{I_2}{I_1} = 1 \times 10^{11}$

A	B
$\beta_1 = 20\text{dB}$	$\beta_2 = 135\text{dB}$ $\beta_2 = 130\text{dB}$

5

$\text{dB} = 10 \log \left(\frac{I}{I_0} \right)$ If a sound has 10 times the sound energy as one with a dB magnitude of 110 on the decibel scale, what is its dB magnitude?

$\frac{I_2}{I_1} = 10$

A	B
$\beta_1 = 110\text{dB}$	$\beta_2 = 120\text{dB}$ $\beta_2 = 114\text{dB}$

6

$\text{pH} = -\log [\text{H}^+]$ If a solution has 1×10^9 times the Hydrogen ion concentration as one with a pH of 11 on the pH scale, what is its pH?

$\frac{[\text{H}^+]_2}{[\text{H}^+]_1} = 1 \times 10^9$

A	B
$\text{pH}_1 = 11$	$\text{pH}_2 = 2$ $\text{pH}_2 = 0.5$

7

$M = \log \left(\frac{I}{I_0} \right)$ If an earthquake has 100 times the wave size as one with a magnitude of 7 on the Richter scale, what is its magnitude?

$\frac{I_2}{I_1} = 100$

A	B
$M_1 = 7$	$M_2 = 9$ $M_2 = 10.5$

8

$\text{pH} = -\log [\text{H}^+]$ If a solution has 100,000 times the Hydrogen ion concentration as one with a pH of 12 on the pH scale, what is its pH?

$\frac{[\text{H}^+]_2}{[\text{H}^+]_1} = 100,000$

A	B
$\text{pH}_1 = 12$	$\text{pH}_2 = 7$ $\text{pH}_2 = 7.5$